

$^{90}\text{Mo}$   $\varepsilon$  decay **1968Co05,1966Pe10**

Type	Author	History	Citation	Literature Cutoff Date
Full Evaluation	S. K. Basu, E. A. Mccutchan		NDS 165, 1 (2020)	1-Mar-2020

Parent:  $^{90}\text{Mo}$ :  $E=0.0$ ;  $J^\pi=0^+$ ;  $T_{1/2}=5.56$  h 9;  $Q(\varepsilon)=2489$  3;  $\% \varepsilon + \% \beta^+$  decay=100.0

**1968Co05**:  $^{93}\text{Nb}(p,4n)$ ,  $E=50-55$  MeV, chemical separation. Measured:  $\gamma$  singles and  $\gamma\gamma$  coin with Ge(Li) and NaI.

**1966Pe10**: measured: ce(t) and  $\beta^+$  spectra.

Others: **1955Ma31**, **1965Co14**, **1965Gr29**, **1981KaZI**, **1984Bu36**.

$\alpha$ : [Additional information 1](#).

 $^{90}\text{Nb}$  Levels

The decay scheme has been established from  $\gamma\gamma$  coin data (**1968Co05**) and excit in  $^{90}\text{Zr}(p,n\gamma)$  (**1972Yo03**).

E(level)	$J^\pi$ †	$T_{1/2}$ †	E(level)	$J^\pi$ †	$T_{1/2}$ †	E(level)	$J^\pi$ †
0	$8^+$	14.60 h 5	382.0 4	$1^+$	6.19 ms 8	1344.1 5	$1^+$
122.370 22	$6^+$	63 $\mu\text{s}$ 2	651.20 21	$4^{(+)}$		1769.1 5	$1^+$
124.7 4	$4^-$	18.91 s 6	822.6 6			1844.8 6	$(1^+)$
285.30 10	$5^+$		827.4 4			2125.6 7	$1^+$
328.00 10	$4^{(+)}$		854.32 23	$2^-$		2309.0 7	$3^+$

† From Adopted Levels.

 $\varepsilon, \beta^+$  radiations

$\varepsilon$  branches are obtained from  $I(\gamma+ce)$  imbalance at each level.

E(decay)	E(level)	$I\beta^+$ †	$I\varepsilon$ †	Log $ft$	$I(\varepsilon+\beta^+)$ †	Comments
(180 3)	2309.0		2.1 6	4.51 13	2.1 6	$\varepsilon\text{K}=0.8492$ 5; $\varepsilon\text{L}=0.1223$ 4; $\varepsilon\text{M}+=0.02846$ 10
(363 3)	2125.6		4.1 5	4.87 6	4.1 5	$\varepsilon\text{K}=0.8619$ 1; $\varepsilon\text{L}=0.11231$ 8; $\varepsilon\text{M}+=0.02580$ 2
(644 3)	1844.8		1.69 24	5.77 7	1.69 24	$\varepsilon\text{K}=0.8668$ ; $\varepsilon\text{L}=0.10841$ 3; $\varepsilon\text{M}+=0.024775$ 7
(720 3)	1769.1		8.0 8	5.20 5	8.0 8	$\varepsilon\text{K}=0.8675$ ; $\varepsilon\text{L}=0.10790$ 2; $\varepsilon\text{M}+=0.024640$ 5
(1145 3)	1344.1		0.73 11	6.65 7	0.73 11	$\varepsilon\text{K}=0.8694$ ; $\varepsilon\text{L}=0.1063$ ; $\varepsilon\text{M}+=0.02422$
(2107 3)	382.0	25 2	56 3	5.30 3	81 5	av $E\beta=477.9$ 14; $\varepsilon\text{K}=0.6020$ 18; $\varepsilon\text{L}=0.07263$ 21; $\varepsilon\text{M}+=0.01652$ 5

† Absolute intensity per 100 decays.

<sup>90</sup>Mo ε decay **1968Co05,1966Pe10** (continued)

γ(<sup>90</sup>Nb)

I<sub>γ</sub> normalization: Calculated from total γ+ce feeding to g.s. assuming no ε feeding to g.s. (0<sup>+</sup> to 8<sup>+</sup> transition).

α(K)exp values given in comments are calculated from I<sub>γ</sub> (1968Co05) and Ice (1966Pe10) assuming α(K)(122.4γ)=0.47 for an E2 transition.

E <sub>γ</sub> <sup>†</sup>	I <sub>γ</sub> <sup>‡@</sup>	E <sub>i</sub> (level)	J <sub>i</sub> <sup>π</sup>	E <sub>f</sub>	J <sub>f</sub> <sup>π</sup>	Mult. <sup>#</sup>	δ	α	I <sub>(γ+ce)</sub> <sup>@</sup>	Comments
(2.3 4)		124.7	4 <sup>-</sup>	122.370	6 <sup>+</sup>	[M2+E3]		3.3×10 <sup>9</sup> 89	8.2 8	ce(M)/(γ+ce)=0.9 11; ce(N)/(γ+ce)=0.10 35; ce(O)/(γ+ce)=4.E-6 13 α(M)=3.0×10 <sup>9</sup> 81; α(N)=3.3×10 <sup>8</sup> 86; α(O)=1.4×10 <sup>4</sup> 17 E <sub>γ</sub> : From level-energy differences. For an attempt to measure the conversion electron energy see 1988GeZV.
42.70 4	2.8 3	328.00	4 <sup>(+)</sup>	285.30	5 <sup>+</sup>	M1+(E2)	<0.18	2.9 4		I <sub>(γ+ce)</sub> : From intensity balance of 124.7 level. α(K)=2.46 22; α(L)=0.38 11; α(M)=0.067 20; α(N)=0.0095 26; α(O)=0.00041 3 Mult.: α(K)exp=2.3 8, K/L=9.5 40 (1966Pe10).
122.370 22	83 3	122.370	6 <sup>+</sup>	0	8 <sup>+</sup>	E2		0.557		δ: From α(K)exp/α(L)exp. α(K)=0.464 7; α(L)=0.0768 11; α(M)=0.01365 20; α(N)=0.00187 3; α(O)=6.63×10 <sup>-5</sup> 10 Mult.: K/L=6.03 18 (1966Pe10).
162.93 9	7.7 7	285.30	5 <sup>+</sup>	122.370	6 <sup>+</sup>	M1+E2	0.24 17	0.067 13		α(K)=0.059 11; α(L)=0.0071 17; α(M)=0.0012 3; α(N)=0.00018 4; α(O)=9.8×10 <sup>-6</sup> 15 Mult.: a(K)exp=0.059 6.
203.13 10	8.2 7	854.32	2 <sup>-</sup>	651.20	4 <sup>(+)</sup>			0.036 4		Mult.: a(K)exp=0.034 4. Mult.: M1+E2 with δ<0.36 is inconsistent with adopted Δπ.
257.34 4	100 3	382.0	1 <sup>+</sup>	124.7	4 <sup>-</sup>	E3(+M4)	<0.12	0.182 12		α(K)=0.149 10; α(L)=0.0269 19; α(M)=0.0048 4; α(N)=0.00066 5; α(O)=2.33×10 <sup>-5</sup> 22 Mult.: a(K)exp=0.150 11, K/L=5.51 16 (1966Pe10).
323.20 18	8.1 7	651.20	4 <sup>(+)</sup>	328.00	4 <sup>(+)</sup>	M1+E2	0.6 3	0.0122 15		δ: From α(K)exp and α(K)exp/α(L)exp. α(K)=0.0107 13; α(L)=0.00126 17; α(M)=0.00022 3; α(N)=3.2×10 <sup>-5</sup> 5; α(O)=1.76×10 <sup>-6</sup> 18 Mult.: a(K)exp=0.0106 12.
<sup>x</sup> 421.0 3	0.32 10									δ: From α(K)exp. a(K)exp=0.0038 14.
425.1 5	0.46 10	1769.1	1 <sup>+</sup>	1344.1	1 <sup>+</sup>					
440.5 <sup>‡</sup> 6	1.2 3	822.6		382.0	1 <sup>+</sup>					

<sup>90</sup>Mo ε decay **1968Co05,1966Pe10** (continued)

γ(<sup>90</sup>Nb) (continued)

<u>E<sub>γ</sub><sup>†</sup></u>	<u>I<sub>γ</sub><sup>‡@</sup></u>	<u>E<sub>i</sub>(level)</u>	<u>J<sub>i</sub><sup>π</sup></u>	<u>E<sub>f</sub></u>	<u>J<sub>f</sub><sup>π</sup></u>	<u>Mult.#</u>	<u>α</u>	<u>Comments</u>
445.37 21	7.7 8	827.4		382.0	1 <sup>+</sup>			Mult.: a(K)exp=0.0050 7.
472.2 3	1.83 19	854.32	2 <sup>-</sup>	382.0	1 <sup>+</sup>			Mult.: a(K)exp=0.0029 4.
489.8 4	0.94 13	1344.1	1 <sup>+</sup>	854.32	2 <sup>-</sup>			Mult.: a(K)exp=0.0038 7.
<sup>x</sup> 517.7 <sup>‡</sup> 7	0.20 13							
941.5 4	7.1 8	1769.1	1 <sup>+</sup>	827.4				Mult.: a(K)exp=0.00071 10.
946.4 <sup>‡</sup> 8	0.9 3	1769.1	1 <sup>+</sup>	822.6				
<sup>x</sup> 987.3 <sup>‡</sup> 10	0.18 7							
990.2 6	1.31 13	1844.8	(1 <sup>+</sup> )	854.32	2 <sup>-</sup>			Mult.: a(K)exp=0.00070 12.
1271.3 6	5.3 5	2125.6	1 <sup>+</sup>	854.32	2 <sup>-</sup>			Mult.: a(K)exp=0.00050 6.
1387.4 5	2.4 3	1769.1	1 <sup>+</sup>	382.0	1 <sup>+</sup>	M1+E2	4.06×10 <sup>-4</sup> 7	α(K)=0.000320 9; α(L)=3.53×10 <sup>-5</sup> 9; α(M)=6.21×10 <sup>-6</sup> 15; α(N)=9.11×10 <sup>-7</sup> 23; α(O)=5.34×10 <sup>-8</sup> 17 Mult.: a(K)exp=0.00044 6.
<sup>x</sup> 1446 <sup>‡</sup> 2	0.06 3							
1454.6 7	2.4 7	2309.0	3 <sup>+</sup>	854.32	2 <sup>-</sup>			Mult.: a(K)exp=0.00036 12.
1463.5 9	0.9 3	1844.8	(1 <sup>+</sup> )	382.0	1 <sup>+</sup>			Mult.: a(K)exp=0.00048 22.
1481.6 14	0.3 3	2309.0	3 <sup>+</sup>	827.4				

<sup>†</sup> From **1966Pe10** (from ce spectra), except as noted.

<sup>‡</sup> From **1968Co05**.

# From conversion coefficient data above and adopted J<sup>π</sup>'s. E3/M2 and MR=0.10 for the other multiplicities. E3/M2 and MR=0.10 for the other multiplicities.

@ For absolute intensity per 100 decays, multiply by 0.774 29.

<sup>x</sup> γ ray not placed in level scheme.

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Legend

- $I_\gamma < 2\% \times I_\gamma^{\text{max}}$
- $I_\gamma < 10\% \times I_\gamma^{\text{max}}$
- $I_\gamma > 10\% \times I_\gamma^{\text{max}}$
- - -  $\gamma$  Decay (Uncertain)
- Coincidence

Decay Scheme

Intensities:  $I_{(\gamma+ce)}$  per 100 parent decays

