

$^{245}\text{Bk } \varepsilon \text{ decay}$     **1976Ah03**

Type	Author	History
Full Evaluation	C. D. Nesaraja	Citation
		NDS 189,1 (2023)

Parent:  $^{245}\text{Bk}$ : E=0.0;  $J^\pi=(3/2^-)$ ;  $T_{1/2}=4.96$  d 3;  $Q(\varepsilon)=809.3$  15; % $\varepsilon$  decay=99.88 1

$^{245}\text{Bk-Q}(\varepsilon)$ : From [2021Wa16](#).

**1976Ah03:**  $^{245}\text{Bk}$  was produced from the irradiation of  $^{243}\text{Am}$  with 34 MeV  $\alpha$  particles at the Argonne 152 cm cyclotron. The chemically mass separated  $^{245}\text{Bk}$  was then processed into a thin source. The electron capture decay of  $^{245}\text{Bk}$  was investigated by measuring the gamma-rays and conversion-electron spectra with Ge(Li) and Si(Li) detectors, respectively. Measured  $T_{1/2}$  of  $^{245}\text{Bk}$ ,  $E\gamma$ ,  $I\gamma$ ,  $E(\text{ce})$ ,  $I(\text{ce})$ ,  $E\gamma$ -coin. Deduced levels, multipolarity,  $J^\pi$  and logft.

**1975Ya03:**  $^{245}\text{Bk}$  was produced from the  $^{241}\text{Am}(\alpha,x)$  reaction at the Argonne 60-in cyclotron. The conversion electron spectra from the chemically separated  $^{245}\text{Bk}$  were measured with a cooled Si(Li) detector. The  $\gamma$  rays were detected with a NaI(Tl) detector. Measured  $T_{1/2}$  of 356-keV level by  $\gamma$ -ce delayed coin.

**1956Ch77:**  $^{245}\text{Bk}$  was produced from the  $^{243}\text{Am}(\alpha,2n)$  and  $^{244}\text{Cm}(\alpha,p2n)$  reactions at the Crocker Laboratory. The gammas were measured with a thallium-activated sodium iodide crystal detector and x-rays were detected using a xenon-filled proportional counter. The electron spectrum in coincidence with K x-rays was observed with an anthracene scintillation crystal. Deduced decay scheme.

**1956Ma32:**  $^{245}\text{Bk}$  was produced from the  $^{244}\text{Cm}(d,n)$  and  $^{243}\text{Am}(\alpha,2n)$  reactions at the Argonne 60-in. cyclotron. Conversion electrons were measured with electron spectrometer, and the gammas were detected with a NaI detector. Measured ratios of conversion electron to K x-rays. Intensities of the x-rays were measured with the NaI detector.

**1979Ah01:** The electron spectrum was measured with a cooled Si(Li) spectrometer and the photon spectrum measured with Ge(Li) diodes. From the measured intensities of K-Auger electrons and K x-rays, K-shell fluorescence yield for Cm was deduced:  $\omega(K)=97.1\%$  6.

Using program RADLST, the evaluator has calculated from the decay scheme, x-ray intensities as follows:  $K\alpha_2$  x ray=35.1 22,  $K\alpha_1$  x ray=55 4,  $K\beta$  x ray=27.9 18. These values are compared with experimental results ([1976Ah03](#)) of  $K\alpha_2$  x ray=35.0 24,  $K\alpha_1$  x ray=54.0 34, and  $(K\beta_1' \times \text{ray}=20.5 15 + K\beta_2' \xi \text{ ray}=7.3 5)= 27.8 16$ . This agreement confirms the quality of the data and the consistency of the decay scheme.

 $^{245}\text{Cm}$  Levels

E(level) <sup>‡</sup>	$J^\pi$ <sup>†</sup>	$T_{1/2}$ <sup>†</sup>	Comments
0.0	$7/2^+$	8423 y 74	
54.819 20	$9/2^+$	$\leq 0.10$ ns	
252.833 23	$5/2^+$		
295.701 17	$7/2^+$		
355.95 10	$1/2^+$	0.29 $\mu\text{s}$ 2	$T_{1/2}$ : From $\gamma$ -ce delayed coincidence method ( <a href="#">1975Ya03</a> ).
361.4 4	$3/2^+$		
418.7 4	$5/2^+$		
633.63 11	$(3/2)^-$		
661.52 8	$(5/2)^-$		
740.97 12	$(1/2^+)$		
769.2 4	$(3/2)^+$		

<sup>†</sup> From Adopted Levels.

<sup>‡</sup> From least-squares fit to  $E\gamma$  data by the evaluator.

**$^{245}\text{Bk } \varepsilon$  decay    1976Ah03 (continued)** $\varepsilon$  radiations

E(decay)	E(level)	I $\varepsilon$ <sup>†‡</sup>	Log ft	Comments
(40.1 16)	769.2	0.184 20	6.42 8	$\varepsilon L=0.38$ 3; $\varepsilon M+=0.62$ 3
(68.3 15)	740.97	1.04 10	6.37 5	$\varepsilon L=0.589$ 5; $\varepsilon M+=0.411$ 5
(147.8 15)	661.52	0.62 6	7.46 5	$\varepsilon K=0.076$ 10; $\varepsilon L=0.633$ 6; $\varepsilon M+=0.291$ 4
(175.7 15)	633.63	2.61 26	7.09 5	$\varepsilon K=0.247$ 9; $\varepsilon L=0.523$ 6; $\varepsilon M+=0.230$ 3
(390.6 16)	418.7			
(447.9 16)	361.4			
(453.4 15)	355.95	5.7 14	8.00 11	$\varepsilon K=0.6659$ 6; $\varepsilon L=0.2420$ 4; $\varepsilon M+=0.09208$ 17 I $\varepsilon$ : Sum of $\varepsilon$ intensities to 355.95, 361.4, 418.7 levels.
(556.5 15)	252.833	90 9	7.02 5	$\varepsilon K=0.6930$ 3; $\varepsilon L=0.2233$ 2; $\varepsilon M+=0.08365$ 10

<sup>†</sup> Deduced by the evaluator from intensity balance at each level.<sup>‡</sup> Absolute intensity per 100 decays.

$^{245}\text{Bk } \varepsilon \text{ decay} \quad \textcolor{blue}{1976Ah03} \text{ (continued)}$  $\gamma(^{245}\text{Cm})$ 

I $\gamma$  normalization: The evaluator has normalized the decay scheme assuming no  $\varepsilon$  feeding neither to the  $^{245}\text{Cm}$  ground state ( $J^\pi=7/2^+$ ) nor to the 55-keV level ( $J^\pi=9/2^+$ ), as expected from  $^{245}\text{Bk}$  ( $J^\pi=3/2^-$ ), and using the sum of the intensities of the 252- and 198-keV  $\gamma$  rays as the full strength of the  $\varepsilon$  decay (99.88 % 1).

## Measured x-ray intensities (1976Ah03)

Energy Intensity x-ray

104.6 1	35.0 24	Cm K $\alpha_2$
109.3 1	54.0 34	Cm K $\alpha_1$
123.2 2	20.5 15	Cm K $\beta_1'$
127.1 2	7.3 5	Cm K $\beta_2'$

E $_\gamma^\dagger$	I $_\gamma^{\dagger b}$	E $_i$ (level)	J $^\pi_i$	E $_f$	J $^\pi_f$	Mult. $^\ddagger$	$\delta$	$\alpha^a$	Comments
(42.87 & 2)		295.701	7/2 $^+$	252.833	5/2 $^+$				
(54.77 & 3)		54.819	9/2 $^+$	0.0	7/2 $^+$				
103.1 1	0.4	355.95	1/2 $^+$	252.833	5/2 $^+$	E2		16.25 24	$\alpha(L)=11.74$ 17; $\alpha(M)=3.32$ 5 $\alpha(N)=0.924$ 14; $\alpha(O)=0.2238$ 33; $\alpha(P)=0.0372$ 5; $\alpha(Q)=0.0001572$ 23 %I $\gamma$ =0.39 L12:L3:M:NO=2.8 2:1.83 13:1.35 13:0.50 6. (1976Ah03); $\alpha(\text{exp})=16.6$ (1975Ya03).
198.0 1	0.17 2	252.833	5/2 $^+$	54.819 9/2 $^+$	E2			1.023 14	I $\gamma$ : Deduced by the evaluator from total intensity provided in Fig. 10 (see note from pg.15) in 1976Ah03 and $\alpha(\text{BrIcc})$ . $\alpha(K)=0.1475$ 21; $\alpha(L)=0.634$ 9; $\alpha(M)=0.1778$ 25 $\alpha(N)=0.0494$ 7; $\alpha(O)=0.01203$ 17; $\alpha(P)=0.002046$ 29; $\alpha(Q)=1.829\times 10^{-5}$ 26 %I $\gamma$ =0.167 23 K:L12:L3= $\leq$ 0.03:0.065 6:0.025 3; $\alpha(K)\text{exp} \leq 0.18$ , $\alpha(L12)\text{exp} = 0.38$ , $\alpha(L3)\text{exp} = 0.15$ (1976Ah03).
(240.90 & 4)		295.701	7/2 $^+$	54.819 9/2 $^+$	M1 $^\#$			2.63 4	$\alpha(K)=2.064$ 29; $\alpha(L)=0.423$ 6; $\alpha(M)=0.1033$ 14 $\alpha(N)=0.0284$ 4; $\alpha(O)=0.00722$ 10; $\alpha(P)=0.001421$ 20; $\alpha(Q)=0.0001015$ 14
252.85 5	31.3 20	252.833	5/2 $^+$	0.0	7/2 $^+$	M1+E2 $^\#$	0.16 @ +6-4	2.25 5	$\alpha(K)=1.76$ 4; $\alpha(L)=0.366$ 6; $\alpha(M)=0.0895$ 14 $\alpha(N)=0.0246$ 4; $\alpha(O)=0.00626$ 10; $\alpha(P)=0.001228$ 20; $\alpha(Q)=8.66\times 10^{-5}$ 21 %I $\gamma$ =30.7 5

**$^{245}\text{Bk } \varepsilon \text{ decay} \quad 1976\text{Ah03 (continued)}$** 

<u><math>\gamma(^{245}\text{Cm})</math> (continued)</u>									
$E_\gamma^{\dagger}$	$I_\gamma^{\dagger b}$	$E_i(\text{level})$	$J_i^\pi$	$E_f$	$J_f^\pi$	Mult. <sup>‡</sup>	$\delta$	$\alpha^a$	Comments
272.2 3	0.013 3	633.63	(3/2) <sup>-</sup>	361.4	3/2 <sup>+</sup>	[E1]		0.0537 8	K:L12:L3:M:NO=53.2 37:11.1 8:0.081 8:2.86 21:1.11 8; $\alpha(K)\text{exp} = 1.70$ , $\alpha(L12)\text{exp} = 0.35$ , $\alpha(L3)\text{exp} = 0.0026$ , $\alpha(M)\text{exp}=0.091$ , $\alpha(NO)\text{exp} = 0.035$ ( <a href="#">1976Ah03</a> ). $\alpha(K)=0.0420$ 6; $\alpha(L)=0.00878$ 12; $\alpha(M)=0.002142$ 30 $\alpha(N)=0.000584$ 8; $\alpha(O)=0.0001458$ 21; $\alpha(P)=2.71\times 10^{-5}$ 4; $\alpha(Q)=1.486\times 10^{-6}$ 21 %I $\gamma=0.0127$ 31
(295.73 <sup>&amp;</sup> 2)		295.701	7/2 <sup>+</sup>	0.0	7/2 <sup>+</sup>	M1+E2 <sup>#</sup>	0.39 <sup>@</sup> +17-24	1.32 14	$\alpha(K)=1.02$ 12; $\alpha(L)=0.223$ 13; $\alpha(M)=0.0550$ 28 $\alpha(N)=0.0151$ 8; $\alpha(O)=0.00384$ 20; $\alpha(P)=0.00075$ 5; $\alpha(Q)=5.0\times 10^{-5}$ 6
350.5 1	0.082 7	769.2	(3/2) <sup>+</sup>	418.7	5/2 <sup>+</sup>	M1		0.929 13	$\alpha(K)=0.731$ 10; $\alpha(L)=0.1487$ 21; $\alpha(M)=0.0363$ 5 $\alpha(N)=0.00996$ 14; $\alpha(O)=0.00254$ 4; $\alpha(P)=0.000499$ 7; $\alpha(Q)=3.56\times 10^{-5}$ 5 %I $\gamma=0.080$ 9
365.8 1	0.39 3	661.52	(5/2) <sup>-</sup>	295.701	7/2 <sup>+</sup>	E1		0.0287 4	K:L12=0.061 6:0.014 2; $\alpha(K)\text{exp} = 0.74$ , $\alpha(L12)\text{exp} = 0.17$ ( <a href="#">1976Ah03</a> ). $\alpha(K)=0.02268$ 32; $\alpha(L)=0.00451$ 6; $\alpha(M)=0.001096$ 15 $\alpha(N)=0.000299$ 4; $\alpha(O)=7.50\times 10^{-5}$ 11; $\alpha(P)=1.412\times 10^{-5}$ 20; $\alpha(Q)=8.27\times 10^{-7}$ 12 %I $\gamma=0.38$ 4
380.8 1	2.58 18	633.63	(3/2) <sup>-</sup>	252.833	5/2 <sup>+</sup>	E1		0.0264 4	Ice(K)=0.013 4; $\alpha(K)\text{exp} = 0.033$ ( <a href="#">1976Ah03</a> ). $\alpha(K)=0.02091$ 29; $\alpha(L)=0.00413$ 6; $\alpha(M)=0.001003$ 14 $\alpha(N)=0.000274$ 4; $\alpha(O)=6.87\times 10^{-5}$ 10; $\alpha(P)=1.296\times 10^{-5}$ 18; $\alpha(Q)=7.65\times 10^{-7}$ 11 %I $\gamma=2.53$ 24
385.0 1	0.61 4	740.97	(1/2 <sup>+</sup> )	355.95	1/2 <sup>+</sup>	M1		0.718 10	L12:M=0.011 1:0.0043 8; $\alpha(L12)\text{exp} = 0.0043$ , $\alpha(M)\text{exp} = 0.0017$ ( <a href="#">1976Ah03</a> ). $\alpha(K)=0.565$ 8; $\alpha(L)=0.1147$ 16; $\alpha(M)=0.0280$ 4 $\alpha(N)=0.00769$ 11; $\alpha(O)=0.001958$ 27; $\alpha(P)=0.000385$ 5; $\alpha(Q)=2.74\times 10^{-5}$ 4 %I $\gamma=0.60$ 6
407.8 2	0.03	769.2	(3/2) <sup>+</sup>	361.4	3/2 <sup>+</sup>				K:L12:M:NO=0.340 25:0.075 7:0.023 2:0.010 2; $\alpha(K)\text{exp} = 0.56$ , $\alpha(L12)\text{exp} = 0.12$ , $\alpha(M)\text{exp} = 0.038$ $\alpha(NO)\text{exp} = 0.016$ ( <a href="#">1976Ah03</a> ). %I $\gamma=0.029$ 6
408.7 1	0.23 2	661.52	(5/2) <sup>-</sup>	252.833	5/2 <sup>+</sup>	[E1]		0.02287 32	ce(K)(407.8 $\gamma$ +408.7 $\gamma$ )=0.017 2 ( <a href="#">1976Ah03</a> ). $\alpha(K)=0.01816$ 25; $\alpha(L)=0.00355$ 5; $\alpha(M)=0.000861$ 12 $\alpha(N)=0.0002349$ 33; $\alpha(O)=5.90\times 10^{-5}$ 8; $\alpha(P)=1.116\times 10^{-5}$ 16; $\alpha(Q)=6.69\times 10^{-7}$ 9 %I $\gamma=0.225$ 25
488.2 2	0.015 3	740.97	(1/2 <sup>+</sup> )	252.833	5/2 <sup>+</sup>	[E2]		0.0623 9	$\alpha(K)=0.0345$ 5; $\alpha(L)=0.02039$ 29; $\alpha(M)=0.00546$ 8

$^{245}\text{Bk } \varepsilon \text{ decay} \quad \textcolor{blue}{1976\text{Ah03}} \text{ (continued)}$  $\gamma(^{245}\text{Cm}) \text{ (continued)}$ 

$E_\gamma^\dagger$	$E_i(\text{level})$	Comments
		$\alpha(N)=0.001511 \ 21; \alpha(O)=0.000373 \ 5; \alpha(P)=6.73\times 10^{-5} \ 9; \alpha(Q)=2.012\times 10^{-6} \ 28$ $\%I\gamma=0.0147 \ 31$

<sup>†</sup> From [1976Ah03](#).

<sup>‡</sup> From experimental conversion-electron data in [1976Ah03](#), except as noted. These values deduced here are given in the Adopted Gammas.

<sup>#</sup> From Adopted Gammas.

<sup>@</sup> From Adopted Gammas.

<sup>&</sup>  $\gamma$ -ray not observed in  $^{245}\text{Bk } \varepsilon$  decay;  $E\gamma$  from Adopted Gammas.

<sup>a</sup> [Additional information 1](#).

<sup>b</sup> For absolute intensity per 100 decays, multiply by 0.98 7.

$^{245}\text{Bk } \varepsilon$  decay    1976Ah03