

<sup>231</sup>U ε decay (4.2 d) 1994Br36,1996Le01

Type	Author	History	Citation	Literature Cutoff Date
Full Evaluation	Balraj Singh, Jagdish K. Tuli, and Edgardo Browne		NDS 185, 560 (2022)	31-Aug-2022

Parent: <sup>231</sup>U: E=0.0; J<sup>π</sup>=(5/2<sup>-</sup>); T<sub>1/2</sub>=4.2 d I; Q(ε)=381.6 20; %ε decay=100.0

<sup>231</sup>U-J<sup>π</sup>,T<sub>1/2</sub>: From <sup>231</sup>U Adopted Levels.

<sup>231</sup>U-Q(ε): From 2021Wa16.

<sup>231</sup>U-%ε decay: %ε≈100 for <sup>231</sup>U ε decay.

1994Br36: <sup>231</sup>U activity was produced via the electron capture decay of <sup>231</sup>Np, which was formed in the <sup>233</sup>U(p,3n) reaction with 49-MeV protons. The recoiling reaction products of <sup>231</sup>Np were chemically separated from their <sup>231</sup>U daughter nuclei. Measured E<sub>γ</sub>, I<sub>γ</sub>, E(ce), I(ce), γγ-coin using HPGe detector for γ rays, pin diode for conversion electrons. Main contaminants were <sup>237</sup>U and <sup>117m</sup>Sn. Deduced conversion coefficients, γ-ray multipolarities.

1996Le01: <sup>231</sup>U activity produced by <sup>232</sup>Th(α,5n), E=52 MeV and mass separated. Measured E<sub>γ</sub>, I<sub>γ</sub> using HPGe detector. Main contaminant was <sup>231</sup>Th, produced in <sup>232</sup>Th(α,α'n) reaction.

1961Ho29, 1957Ho07: data from private communications from J.M. Hollander listed in 1978LeZA (Table of Isotopes).

Evaluators' note: the decay scheme seems fairly complete, except for lack of spectral information for 9.18 and 77.69-keV γ transitions.

<sup>231</sup>Pa Levels

E(level) <sup>†</sup>	J <sup>π</sup> <sup>‡</sup>	T <sub>1/2</sub> <sup>‡</sup>	Comments
0.0	3/2 <sup>-</sup>	32570 y 130	
9.186 20	1/2 <sup>-</sup>		
58.570 3	7/2 <sup>-</sup>	274 ps 10	
77.69 4	5/2 <sup>-</sup>		
84.2150 20	5/2 <sup>+</sup>	45.1 ns 13	
101.409 4	7/2 <sup>+</sup>	0.7 ns 2	T <sub>1/2</sub> : combined for 101.4+102.3 levels (1975Ho14, <sup>231</sup> Th β <sup>-</sup> decay).
102.2687 22	3/2 <sup>+</sup>	0.7 ns 2	T <sub>1/2</sub> : combined for 101.4+102.3 levels (1975Ho14, <sup>231</sup> Th β <sup>-</sup> decay).
174.160 5	(5/2 <sup>-</sup> )		
183.496 3	5/2 <sup>+</sup>	≤0.19 ns	
273.719 16	(1/2 <sup>+</sup> )		
317.94 5	(3/2 <sup>+</sup> )	0.07 ns +11-3	
320.207 22	3/2 <sup>-</sup>		

<sup>†</sup> Deduced from least-squares fit to E<sub>γ</sub> values.

<sup>‡</sup> From the Adopted Levels.

ε radiations

E(decay)	E(level)	Iε <sup>†</sup>	Log ft	Comments
(61.4 20)	320.207	0.53 10	6.3 1	εL=0.603 7; εM+=0.397 7
(63.7 20)	317.94	0.0028 3	8.6 1	εL=0.610 7; εM+=0.390 7
(107.9 20)	273.719	0.13 3	7.5 <sup>1u</sup> 1	εL=0.6777 16; εM+=0.3223 16
(198.1 20)	183.496	0.14 7	8.4 2	εK=0.460 7; εL=0.384 5; εM+=0.1554 20
(207.4 20)	174.160	0.06 3	8.8 2	εK=0.487 6; εL=0.366 4; εM+=0.1469 18
(279.3 20)	102.2687	59 23	6.2 2	εK=0.6073 21; εL=0.2837 15; εM+=0.1089 7
(297.4 20)	84.2150	44 23	6.4 2	εK=0.6245 18; εL=0.2719 12; εM+=0.1036 6
(303.9 20)	77.69	0.52 8	8.4 1	εK=0.6300 17; εL=0.2681 12; εM+=0.1019 5

<sup>†</sup> Absolute intensity per 100 decays.

γ(<sup>231</sup>Pa)

I<sub>γ</sub> normalization: From a relative K x ray intensity of 780 28 (1994Br36), a theoretical εK/ε=0.59 (1977Ba48), and a K-fluorescence yield of 0.970 4 (1996Sc06) the total number of atomic vacancies produced by the electron capture process and the I<sub>γ</sub> normalization factor become 1362 66 and 0.073 4, respectively. The total number of atomic vacancies also agrees with the total relative γ-ray transition intensity (photons plus electrons) of 1350 70 to the ground and 9.2-keV levels. This result is consistent with the absence of any appreciable electron-capture decays to these levels.

Measured energies and intensities of Pa x-rays relative to 100 for the 84.21-keV γ ray from <sup>231</sup>U ε decay (1994Br36):

Pa K<sub>α2</sub>: 92.3 keV 1, relative intensity=238 14 (1994Br36).

Pa K<sub>α1</sub>: 95.8 keV 1, relative intensity=382 22 (1994Br36).

Pa K<sub>β3</sub>: 107.5 keV 2, relative intensity=34.6 28 (1994Br36).

Pa K<sub>β1</sub>: 108.3 keV 2, relative intensity=78 6 (1994Br36).

Pa K<sub>β2</sub>: 111.4 keV 2, relative intensity=37.6 26 (1994Br36).

Pa O<sub>23</sub>: 112.2 keV 2, relative intensity=9.2 12 (1994Br36).

Experimental Pa K x ray intensities: 17.5% 10 (Kα<sub>2</sub> x ray), 28.1% 16 (Kα<sub>1</sub> x ray), and 11.8% 6 (Kβ x ray) compare with 17.85% 8 (Kα<sub>2</sub> x ray), 28.73% 12 (Kβ x ray), respectively, calculated by evaluators (using the computer program RADLST) from γ-ray intensities, K-conversion coefficients and K-electron capture probabilities presented here, using a K-fluorescence yield of 0.970 4 (1996Sc06). This agreement shows that most γ rays with energies greater than 112.6 keV (the K-binding in Pa) have accurate intensities and correct multiplicities, and that K-electron capture probabilities are also correct.

Q(β<sup>-</sup>)=381.6 keV 20 (2021Wa16) compares with a total average radiation energy of 350 keV, calculated by evaluators using the computer program RADLST. The small discrepancy between these values is probably due to low-energy γ rays (i.e., 9.2-, 17.2-, and 19.1-keV transitions) not included in the calculation because their intensities are unknown. These results support the consistency and completeness of the decay scheme.

<u>E<sub>γ</sub><sup>†</sup></u>	<u>I<sub>γ</sub><sup>‡b</sup></u>	<u>E<sub>i</sub>(level)</u>	<u>J<sub>i</sub><sup>π</sup></u>	<u>E<sub>f</sub></u>	<u>J<sub>f</sub><sup>π</sup></u>	<u>Mult.<sup>a</sup></u>	<u>δ<sup>a</sup></u>	<u>α<sup>c</sup></u>	<u>I<sub>(γ+ce)</sub><sup>b</sup></u>	<u>Comments</u>
(9.183)		9.186	1/2 <sup>-</sup>	0.0	3/2 <sup>-</sup>				8.8 3	E <sub>γ</sub> : from the Adopted Gammas. I <sub>(γ+ce)</sub> : from transition intensity balance at 9.19 level, assuming no direct ε feeding to the 9.19 level, as expected from ΔJ <sup>π</sup> =(2), Δπ=(no). %I <sub>γ</sub> ≤0.0010 α(L)=1.4; α(M)=122.6 18; α(N)=32.7 5; α(O)=7.75 12; α(P)=1.505 22; α(Q)=0.144 2 I <sub>(γ+ce)</sub> : from I(γ+ce)≤2.3 for 82.087γ from 183.4 level. I <sub>γ</sub> : from I(γ+ce) and α(theory). %I <sub>γ</sub> =0.26 10 α(L)=45 15; α(M)=128 8 α(N)=34 2; α(O)=8.1 4; α(P)=1.55 5; α(Q)=0.124 1 %I <sub>γ</sub> =14.6 12 M1/M3=0.38; M2/M3=0.61; M4/M3=0.38; M5/M3=0.38 (1961Ho29)
(17.195 21)	≤0.014	101.409	7/2 <sup>+</sup>	84.2150	5/2 <sup>+</sup>	(M1)			166 3	≤2.3
(18.055 18)	3.6 <sup>&amp;</sup> 13	102.2687	3/2 <sup>+</sup>	84.2150	5/2 <sup>+</sup>	M1+E2	0.040 +18-0		218 28	
25.65 2	200 <sup>@</sup> 12	84.2150	5/2 <sup>+</sup>	58.570	7/2 <sup>-</sup>	E1			4.37 6	

<sup>231</sup>U ε decay (4.2 d) [1994Br36,1996Le01](#) (continued)

γ(<sup>231</sup>Pa) (continued)

<u>E<sub>γ</sub><sup>†</sup></u>	<u>I<sub>γ</sub><sup>‡b</sup></u>	<u>E<sub>i</sub>(level)</u>	<u>J<sub>i</sub><sup>π</sup></u>	<u>E<sub>f</sub></u>	<u>J<sub>f</sub><sup>π</sup></u>	<u>Mult.<sup>a</sup></u>	<u>δ<sup>a</sup></u>	<u>α<sup>c</sup></u>	<u>Comments</u>
58.5700 24	6.8 <sup>@</sup> 4	58.570	7/2 <sup>-</sup>	0.0	3/2 <sup>-</sup>	E2		155.5 22	α(L)=3.25 5; α(M)=0.842 13 α(N)=0.219 4; α(O)=0.0471 7; α(P)=0.00673 10; α(Q)=0.000196 3 E <sub>γ</sub> : other: 25.65 4 ( <a href="#">1994Br36</a> ). %I <sub>γ</sub> =0.50 4 α(L1)exp+α(L2)exp=64 8; α(L3)exp=49 6; α(M)exp=36 5; α(N)exp=10 2 ( <a href="#">1994Br36</a> ) α(L)=113.6 16; α(M)=31.3 5 α(N)=8.43 12; α(O)=1.90 3; α(P)=0.306 5; α(Q)=0.000818 12 E <sub>γ</sub> : other: 58.55 4 ( <a href="#">1994Br36</a> ). I <sub>γ</sub> : 6.4 4 in <a href="#">1994Br36</a> is adjusted by evaluators to 6.8 4 to obtain transition intensity balance at 58.57 level, with no direct ε feeding to this level. Ice(L1+L2)=205 21, Ice(L3)=156 16, Ice(M)=116 12, Ice(N)=31 3 ( <a href="#">1994Br36</a> ). Other: <a href="#">1961Ho29</a> .
68.5 1	0.079	77.69	5/2 <sup>-</sup>	9.186	1/2 <sup>-</sup>	E2		73.3 10	<a href="#">Additional information 1</a> . %I <sub>γ</sub> =0.00577 α(L)=53.5 8; α(M)=14.76 21 α(N)=3.98 6; α(O)=0.898 13; α(P)=0.1448 21; α(Q)=0.000423 6 <a href="#">Additional information 2</a> . I <sub>γ</sub> : from <a href="#">1961Ho29</a> , cited by <a href="#">1994Br36</a> , where I <sub>γ</sub> is deduced from relative conversion electron intensity and α(theory) for E2.
(72.7510 29)	0.15 <sup>&amp;</sup> 8	174.160	(5/2 <sup>-</sup> )	101.409	7/2 <sup>+</sup>	[E1]		0.280 4	%I <sub>γ</sub> =0.011 6 α(L)=0.211 3; α(M)=0.0517 8 α(N)=0.01363 20; α(O)=0.00310 5; α(P)=0.000519 8; α(Q)=2.33×10 <sup>-5</sup> 4
(77.69)	0.057	77.69	5/2 <sup>-</sup>	0.0	3/2 <sup>-</sup>	[M1+E2]		24 16	%I <sub>γ</sub> =0.00416 α(L)=18 12; α(M)=5 4 α(N)=1.3 9; α(O)=0.30 20; α(P)=0.05 3; α(Q)=0.0009 7 E <sub>γ</sub> : from level-energy difference.
81.2280 21	≤0.32	183.496	5/2 <sup>+</sup>	102.2687	3/2 <sup>+</sup>	M1(+E2)	0.00 8	7.66 20	%I <sub>γ</sub> ≤0.0234 α(L)=5.78 14; α(M)=1.40 4 α(N)=0.374 11; α(O)=0.0898 24; α(P)=0.0172 4; α(Q)=0.001422 22 I <sub>γ</sub> : from <a href="#">1996Le01</a> . Other: I <sub>γ</sub> =0.19, Ice(L1)=0.51 ( <a href="#">1961Ho29</a> , cited by <a href="#">1994Br36</a> ).
82.0870 22	≤0.26	183.496	5/2 <sup>+</sup>	101.409	7/2 <sup>+</sup>	M1(+E2)	0.04 6	7.47 23	%I <sub>γ</sub> ≤0.019 α(L)=5.63 17; α(M)=1.36 5 α(N)=0.365 12; α(O)=0.088 3; α(P)=0.0167 5; α(Q)=0.001377 22 I <sub>γ</sub> : from <a href="#">1996Le01</a> . Other: I <sub>γ</sub> =0.26, Ice(L1)=0.73 ( <a href="#">1961Ho29</a> , cited by <a href="#">1994Br36</a> ).

<sup>231</sup>U ε decay (4.2 d) [1994Br36](#),[1996Le01](#) (continued)

γ(<sup>231</sup>Pa) (continued)

$E_\gamma$ †	$I_\gamma$ ‡b	$E_i$ (level)	$J_i^\pi$	$E_f$	$J_f^\pi$	Mult. <sup>a</sup>	$\delta^a$	$\alpha^c$	Comments
84.2140 22	100@ 6	84.2150	5/2 <sup>+</sup>	0.0	3/2 <sup>-</sup>	E1		2.17 21	%I <sub>γ</sub> =7.3 6 α(L1)exp+α(L2)exp=1.7 1; α(M)exp=0.48 4; α(N)exp=0.34 3 ( <a href="#">1994Br36</a> ) α(M1)exp=0.34; α(M2)exp=0.21; α(M3)exp=0.009; α(N1)exp=0.17 ( <a href="#">1961Ho29</a> ) E <sub>γ</sub> : other: 84.23 7 ( <a href="#">1994Br36</a> ). Ice(L1+L2)=83 1, Ice(M)=24 1, Ice(N)=17 1 ( <a href="#">1994Br36</a> ). Other: <a href="#">1961Ho29</a> . <a href="#">Additional information 3</a> . Total conversion coefficient from ce data in <sup>231</sup> Th β <sup>-</sup> decay. Other: 2.6 1 from summed conversion coefficients in <a href="#">1994Br36</a> . Transition is anomalously converted ( <a href="#">1960As02</a> ). See also <a href="#">2008Go10</a> . α(O1)exp=0.031 ( <a href="#">1961Ho29</a> ).
89.95 2	0.58# 32	174.160	(5/2 <sup>-</sup> )	84.2150	5/2 <sup>+</sup>	(E1)		0.1598 23	%I <sub>γ</sub> =0.042 23 α(L)=0.1205 17; α(M)=0.0294 5 α(N)=0.00777 11; α(O)=0.001782 25; α(P)=0.000304 5; α(Q)=1.467×10 <sup>-5</sup> 21
(93.02 4)	1.8& 2	102.2687	3/2 <sup>+</sup>	9.186	1/2 <sup>-</sup>	(E1)		0.1463 21	%I <sub>γ</sub> =0.131 16 α(L)=0.1103 16; α(M)=0.0269 4 α(N)=0.00711 10; α(O)=0.001633 23; α(P)=0.000279 4; α(Q)=1.363×10 <sup>-5</sup> 20
(99.2780 36)	≤0.050&	183.496	5/2 <sup>+</sup>	84.2150	5/2 <sup>+</sup>	M1+E2	0.35 7	5.2 4	%I <sub>γ</sub> ≤0.00365 α(L)=3.9 3; α(M)=0.97 8 α(N)=0.261 20; α(O)=0.062 5; α(P)=0.0113 7; α(Q)=0.00072 3
102.2700 24	17.2@ 10	102.2687	3/2 <sup>+</sup>	0.0	3/2 <sup>-</sup>	(E1)		0.1141 16	%I <sub>γ</sub> =1.26 10 α(L)=0.0860 13; α(M)=0.0209 3 α(N)=0.00554 8; α(O)=0.001275 19; α(P)=0.000220 4; α(Q)=1.106×10 <sup>-5</sup> 16 E <sub>γ</sub> : other: 102.3 1 ( <a href="#">1994Br36</a> ).
136.75 7	0.57# 5	320.207	3/2 <sup>-</sup>	183.496	5/2 <sup>+</sup>	[E1]		0.237 3	%I <sub>γ</sub> =0.042 4 α(K)=0.184 3; α(L)=0.0404 6; α(M)=0.00981 14 α(N)=0.00260 4; α(O)=0.000603 9; α(P)=0.0001061 15; α(Q)=5.83×10 <sup>-6</sup> 9
171.43# 3	0.200# 16	273.719	(1/2 <sup>+</sup> )	102.2687	3/2 <sup>+</sup>	[M1+E2]		2.8 16	%I <sub>γ</sub> =0.0146 14 α(K)=1.8 17; α(L)=0.72 6; α(M)=0.19 3 α(N)=0.051 7; α(O)=0.0118 14; α(P)=0.00207 8; α(Q)=9.E-5 8
(189.5)	≤0.0053#	273.719	(1/2 <sup>+</sup> )	84.2150	5/2 <sup>+</sup>				%I <sub>γ</sub> ≤0.00039 E <sub>γ</sub> : from level-energy difference.

<sup>231</sup>U ε decay (4.2 d) [1994Br36](#),[1996Le01](#) (continued)

γ(<sup>231</sup>Pa) (continued)

$E_\gamma$ †	$I_\gamma$ ‡ <sup>b</sup>	$E_i$ (level)	$J_i^\pi$	$E_f$	$J_f^\pi$	Mult. <sup>a</sup>	$\delta^a$	$\alpha^c$	Comments
(196.0)	≤0.0053 <sup>#</sup>	273.719	(1/2 <sup>+</sup> )	77.69	5/2 <sup>-</sup>				%I <sub>γ</sub> ≤0.00039 γ to 5/2 <sup>-</sup> is unlikely. E <sub>γ</sub> : from level-energy difference.
217.94 3	4.1 11	320.207	3/2 <sup>-</sup>	102.2687	3/2 <sup>+</sup>	[E1]		0.0789 11	%I <sub>γ</sub> =0.30 8 α(K)=0.0624 9; α(L)=0.01249 18; α(M)=0.00301 5 α(N)=0.000801 12; α(O)=0.000188 3; α(P)=3.39×10 <sup>-5</sup> 5; α(Q)=2.10×10 <sup>-6</sup> 3 E <sub>γ</sub> : other: 217.9 2 ( <a href="#">1994Br36</a> ). Energy is from <a href="#">1994Br36</a> . I <sub>γ</sub> is a unweighted average of discrepant values: 5.26 3 ( <a href="#">1996Le01</a> ) and 3.0 4 ( <a href="#">1994Br36</a> ).
236.04 6	1.22 <sup>#</sup> 6	320.207	3/2 <sup>-</sup>	84.2150	5/2 <sup>+</sup>	[E1]		0.0657 9	%I <sub>γ</sub> =0.089 7 α(K)=0.0521 8; α(L)=0.01028 15; α(M)=0.00248 4 α(N)=0.000659 10; α(O)=0.0001545 22; α(P)=2.80×10 <sup>-5</sup> 4; α(Q)=1.770×10 <sup>-6</sup> 25
240.27 5	0.0132 <sup>#</sup> 16	317.94	(3/2 <sup>+</sup> )	77.69	5/2 <sup>-</sup>	[E1]		0.0630 9	%I <sub>γ</sub> =0.00096 13 α(K)=0.0500 7; α(L)=0.00984 14; α(M)=0.00237 4 α(N)=0.000631 9; α(O)=0.0001480 21; α(P)=2.68×10 <sup>-5</sup> 4; α(Q)=1.703×10 <sup>-6</sup> 24 E <sub>γ</sub> : other: 240.3 2 ( <a href="#">1996Le01</a> ).
242.51 4	0.111 <sup>#</sup> 11	320.207	3/2 <sup>-</sup>	77.69	5/2 <sup>-</sup>	[M1+E2]		1.01 66	%I <sub>γ</sub> =0.0081 9 α(K)=0.72 61; α(L)=0.22 4; α(M)=0.055 7 α(N)=0.0147 17; α(O)=0.0035 5; α(P)=0.00062 13; α(Q)=3.5×10 <sup>-5</sup> 28
264.55 <sup>#</sup> 2	0.279 <sup>#</sup> 26	273.719	(1/2 <sup>+</sup> )	9.186	1/2 <sup>-</sup>	[E1]		0.0506 7	%I <sub>γ</sub> =0.0204 22 α(K)=0.0403 6; α(L)=0.00781 11; α(M)=0.00188 3 α(N)=0.000500 7; α(O)=0.0001176 17; α(P)=2.14×10 <sup>-5</sup> 3; α(Q)=1.389×10 <sup>-6</sup> 20
273.71 <sup>#</sup> 2	0.621 <sup>#</sup> 32	273.719	(1/2 <sup>+</sup> )	0.0	3/2 <sup>-</sup>	[E1]		0.0469 7	%I <sub>γ</sub> =0.0453 34 α(K)=0.0373 6; α(L)=0.00721 10; α(M)=0.001735 25 α(N)=0.000461 7; α(O)=0.0001085 16; α(P)=1.98×10 <sup>-5</sup> 3; α(Q)=1.292×10 <sup>-6</sup> 18
308.78 7	0.0195 <sup>#</sup> 21	317.94	(3/2 <sup>+</sup> )	9.186	1/2 <sup>-</sup>	[E1]		0.0358 5	%I <sub>γ</sub> =0.00142 17 α(K)=0.0287 4; α(L)=0.00544 8; α(M)=0.001307 19 α(N)=0.000348 5; α(O)=8.19×10 <sup>-5</sup> 12; α(P)=1.501×10 <sup>-5</sup> 22; α(Q)=1.006×10 <sup>-6</sup> 15 E <sub>γ</sub> : other: 308.7 2 ( <a href="#">1996Le01</a> ).
311.00 5	0.37 <sup>#</sup> 5	320.207	3/2 <sup>-</sup>	9.186	1/2 <sup>-</sup>	M1(+E2)	0.7 9	0.61 27	%I <sub>γ</sub> =0.027 4 α(K)=0.47 24; α(L)=0.107 24; α(M)=0.026 5 α(N)=0.0071 14; α(O)=0.0017 4; α(P)=0.00031 8; α(Q)=2.2×10 <sup>-5</sup> 11

<sup>231</sup>U  $\epsilon$  decay (4.2 d) [1994Br36](#),[1996Le01](#) (continued)

$\gamma(^{231}\text{Pa})$  (continued)

$E_\gamma^\dagger$	$I_\gamma^{\ddagger b}$	$E_i(\text{level})$	$J_i^\pi$	$E_f$	$J_f^\pi$	Mult. <sup>a</sup>	$\alpha^c$	Comments
317.87 8	0.0042 <sup>#</sup> 16	317.94	(3/2 <sup>+</sup> )	0.0	3/2 <sup>-</sup>	[E1]	0.0336 5	%I $\gamma$ =0.00031 12 $\alpha(\text{K})=0.0269$ 4; $\alpha(\text{L})=0.00509$ 8; $\alpha(\text{M})=0.001223$ 18 $\alpha(\text{N})=0.000325$ 5; $\alpha(\text{O})=7.67\times 10^{-5}$ 11; $\alpha(\text{P})=1.406\times 10^{-5}$ 20; $\alpha(\text{Q})=9.48\times 10^{-7}$ 14 $E_\gamma$ : other: 317.7 3 ( <a href="#">1996Le01</a> ).
320.15 8	0.0158 16	320.207	3/2 <sup>-</sup>	0.0	3/2 <sup>-</sup>	[M1+E2]	0.46 32	%I $\gamma$ =0.00115 13 $\alpha(\text{K})=0.34$ 28; $\alpha(\text{L})=0.088$ 29; $\alpha(\text{M})=0.0221$ 61 $\alpha(\text{N})=0.0059$ 17; $\alpha(\text{O})=0.00140$ 41; $\alpha(\text{P})=2.57\times 10^{-4}$ 89; $\alpha(\text{Q})=1.6\times 10^{-5}$ 13

<sup>†</sup> From the Adopted Gammas, which are mostly from <sup>231</sup>Th  $\beta^-$  decay, unless otherwise indicated.

<sup>‡</sup> Relative intensities listed in [1994Br36](#) and [1996Le01](#) are normalized to 100 for the 84.21-keV  $\gamma$  ray from a level of this energy.

<sup>#</sup> From [1996Le01](#).

<sup>@</sup> From [1994Br36](#).

<sup>&</sup> Deduced from the  $\gamma$ -branching ratios in Adopted Gammas, which are mostly from <sup>231</sup>Th  $\beta^-$  decay.

<sup>a</sup> From the Adopted Gammas, unless otherwise stated.

<sup>b</sup> For absolute intensity per 100 decays, multiply by 0.073 4.

<sup>c</sup> Total theoretical internal conversion coefficients, calculated using the BrIcc code ([2008Ki07](#)) with Frozen orbital approximation based on  $\gamma$ -ray energies, assigned multipolarities, and mixing ratios, unless otherwise specified.

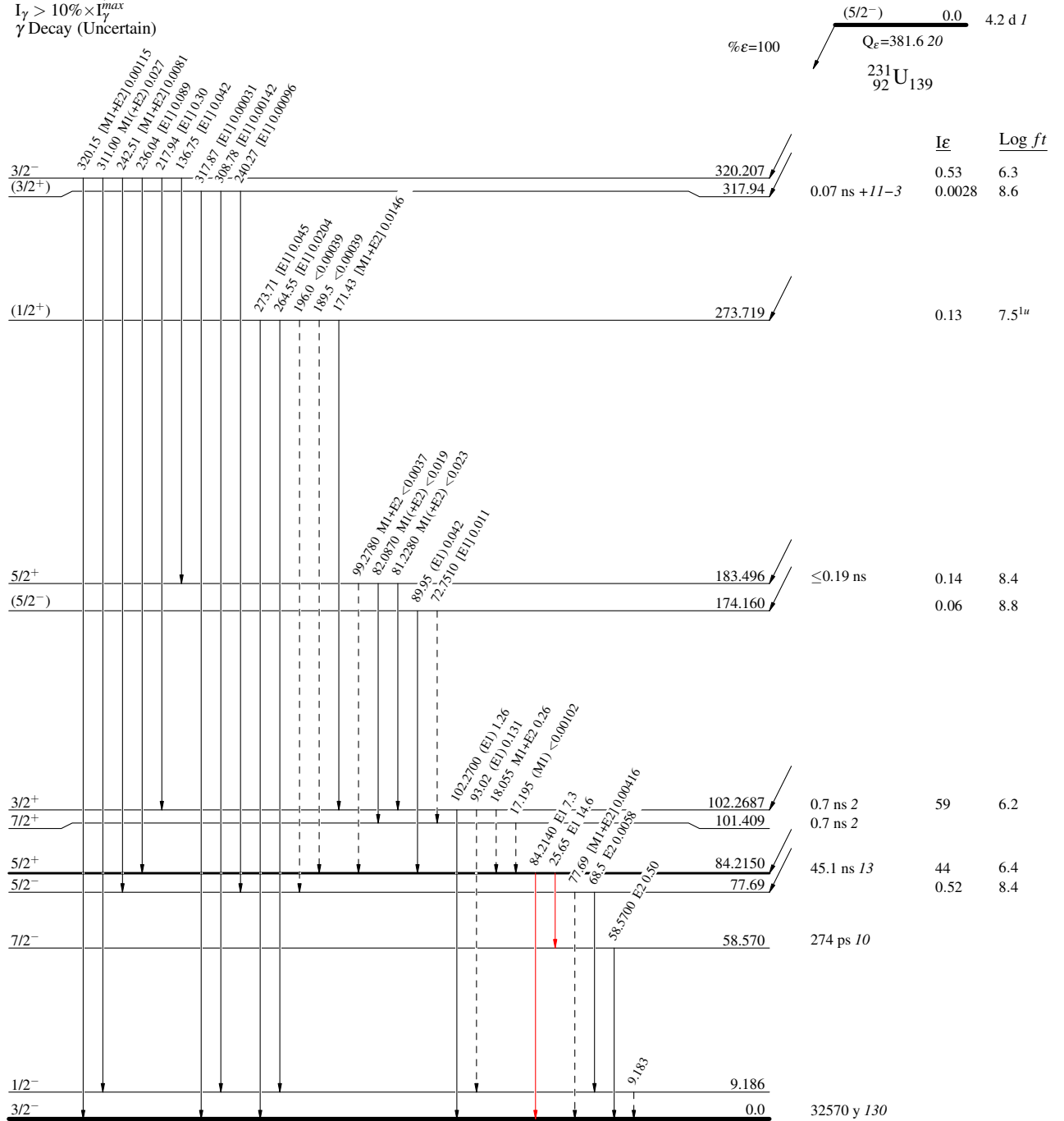
$^{231}\text{U}$   $\epsilon$  decay (4.2 d) 1994Br36,1996Le01

Decay Scheme

Legend

- $I_\gamma < 2\% \times I_\gamma^{max}$
- $I_\gamma < 10\% \times I_\gamma^{max}$
- $I_\gamma > 10\% \times I_\gamma^{max}$
- - -  $\gamma$  Decay (Uncertain)

Intensities:  $I_\gamma$  per 100 parent decays



$^{231}\text{Pa}_{140}$