

$^{196}\text{Ir } \beta^-$  decay (1.40 h) 1968Ja06

Type	Author	History
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		NDS 108, 1093 (2007)

Parent:  $^{196}\text{Ir}$ :  $E=4.1\times 10^2$  11;  $J^\pi=(10,11^-)$ ;  $T_{1/2}=1.40$  h 2;  $Q(\beta^-)=3209$  38; % $\beta^-$  decay=100.0 $^{196}\text{Ir}$ -% $\beta^-$  decay: IT decay<0.3% from absence of 779-keV  $\gamma$  present in decay of 52-s  $^{196}\text{Ir}$ .

Also 1965Bi04, 1967JaZZ.

1968Ja06, 1970To14: source prepared by  $^{198}\text{Pt}(d,\alpha)$  and  $^{196}\text{Pt}(n,p)$ , enriched, chem,  $\gamma$ , ce; semi. $^{196}\text{Pt}$  Levels

E(level) <sup>†</sup>	$J^\pi$ <sup>‡</sup>	$T_{1/2}$ <sup>#</sup>	Comments
0.0	0 <sup>+</sup>	stable	
355.90 20	2 <sup>+</sup>		
877.2 3	4 <sup>+</sup>		
1270.7 3	5 <sup>-</sup>	1.1 ns 2	$J^\pi$ : supported by E1 deexcitation to 877 level. $J^\pi$ : also log $ft$ from $^{196}\text{Au}$ $\varepsilon$ decay.
1374.0 4	7 <sup>-</sup>	4.01 ns 16	$T_{1/2}$ : from delayed coin (1970To14); see also 1968Ja06.
1430.1? 4	(5,6 <sup>+</sup> )		$J^\pi$ : based upon E2 deexcitation to 5 <sup>-</sup> level.
1821.1 4	9 <sup>-</sup>	<1 ns	$T_{1/2}$ : from delayed coin (1970To14); see also 1968Ja06.
2161.9? 5	(9 <sup>-</sup> ,10,11 <sup>-</sup> )		$J^\pi$ : $\gamma$ 's to 9 <sup>-</sup> , $\gamma$ 's from (9 <sup>-</sup> ,10,11 <sup>-</sup> ).
2454.6 4	7 <sup>-</sup> ,8		$J^\pi$ : see 1430 level.
2468.4 4	10 <sup>-</sup> ,11 <sup>-</sup>	<1 ns	$J^\pi$ : based upon E2 deexcitation to 9 <sup>-</sup> level, and log $ft$ =6.05, no $\gamma$ 's to $J^\pi<9$ . $T_{1/2}$ : from $\beta\gamma(t)$ measurements (1968Ja06).
2889.3? 5	(9 <sup>-</sup> ,10,11 <sup>-</sup> )		$J^\pi$ : $\gamma$ 's to 11 <sup>-</sup> and 9 <sup>-</sup> , log $ft$ =6.5 from (10,11 <sup>-</sup> ).
3162.4 5	(9 <sup>-</sup> ,10,11 <sup>-</sup> )		$J^\pi$ : $\gamma$ 's to 11 <sup>-</sup> and 9 <sup>-</sup> , log $ft$ =5.9 from (10,11 <sup>-</sup> ).
3176.8? 5	(9 <sup>-</sup> )		$J^\pi$ : $\gamma$ 's to 7 <sup>-</sup> and 9 <sup>-</sup> , log $ft$ =6.7 from (10,11 <sup>-</sup> ).
3215.2? 5	(9 <sup>-</sup> )		$J^\pi$ : $\gamma$ 's to 7 <sup>-</sup> and 9 <sup>-</sup> , log $ft$ =6.5 from (10,11 <sup>-</sup> ).
3304.0 4	(10,11 <sup>-</sup> )		$J^\pi$ : $\gamma$ 's to 11 <sup>-</sup> and 9 <sup>-</sup> , log $ft$ =5.1 from (10,11 <sup>-</sup> ).

<sup>†</sup> From least-squares fit to  $E\gamma$ 's.<sup>‡</sup> From the Adopted Levels. Contributing arguments from this data set are given as comments.

# From the Adopted Levels, except as noted.

 $\beta^-$  radiationsEvaluation of transition balance at each level gives no evidence of  $\beta^-$  feeding below the 2468 level.

E(decay)	E(level)	$I\beta^-$ <sup>†</sup>	Log $ft$	Comments
(3.2×10 <sup>2</sup> 12)	3304.0	9.1 4	5.1 7	av $E\beta=89$ 38
(4.0×10 <sup>2</sup> 12)	3215.2?	0.82 6	6.5 5	av $E\beta=117$ 39
(4.4×10 <sup>2</sup> 12)	3176.8?	0.70 8	6.7 5	av $E\beta=130$ 40
(4.6×10 <sup>2</sup> 12)	3162.4	4.5 4	5.9 5	av $E\beta=134$ 40
(7.3×10 <sup>2</sup> 12)	2889.3?	5.2 2	6.5 3	av $E\beta=230$ 43
1.16×10 <sup>3</sup> 10	2468.4	80 4	6.03 17	av $E\beta=390$ 47

<sup>†</sup> Absolute intensity per 100 decays.

<sup>196</sup>Ir  $\beta^-$  decay (1.40 h)    **1968Ja06 (continued)** $\gamma(^{196}\text{Pt})$ I $\gamma$  normalization: Based upon I( $\gamma$ +ce)=100 for the 356-keV transition.

E $_{\gamma}$	I $_{\gamma}$ <sup>‡&amp;</sup>	E $_i$ (level)	J $^{\pi}_i$	E $_f$	J $^{\pi}_f$	Mult. <sup>†</sup>	$\alpha^a$	Comments
103.3 2	19.2 22	1374.0	7 <sup>-</sup>	1270.7	5 <sup>-</sup>	E2	4.28 7	$\alpha(K)=0.685$ 10; $\alpha(L)=2.70$ 5; $\alpha(M)=0.699$ 12; $\alpha(N+..)=0.197$ 4 Mult.: supported by $\alpha(L)\exp=2.8$ 5, L/M+=3.9 5; I $\gamma$ : Required for intensity balance at 1271 and 1374 level. Measured I $\gamma$ : 17 2 (1968Ja06).
<sup>x</sup> 333.0 <sup>@</sup>	<0.25							
340.7 <sup>#</sup> 4	1.6 2	2161.9?	(9 <sup>-</sup> ,10,11 <sup>-</sup> )	1821.1	9 <sup>-</sup>			$\alpha(K)=0.0402$ 6; $\alpha(L)=0.01516$ 22; $\alpha(M)=0.00376$ 6; $\alpha(N+..)=0.001079$ 16
355.9 2	98 3	355.90	2 <sup>+</sup>	0.0	0 <sup>+</sup>	E2	0.0602	Mult.: supported by K/L=2.1 1, L/M+=2.9 7. Mult.: see also <sup>196</sup> Au $\varepsilon$ decay and <sup>196</sup> Pt Coulomb excitation.
393.5 2	101 2	1270.7	5 <sup>-</sup>	877.2	4 <sup>+</sup>	E1	0.01394	$\alpha(K)=0.01158$ 17; $\alpha(L)=0.00182$ 3; $\alpha(M)=0.000418$ 6; $\alpha(N+..)=0.0001219$ 18 Mult.: supported by $\alpha(K)\exp=0.012$ 2, K/L=5.3 2.
420.9 <sup>#</sup> 3	2.6 1	2889.3?	(9 <sup>-</sup> ,10,11 <sup>-</sup> )	2468.4	10 <sup>-</sup> ,11 <sup>-</sup>			$\alpha(K)=0.0235$ 4; $\alpha(L)=0.00702$ 10; $\alpha(M)=0.001717$ 25; $\alpha(N+..)=0.000495$ 7
447.1 2	98 2	1821.1	9 <sup>-</sup>	1374.0	7 <sup>-</sup>	E2	0.0327	Mult.: supported by $\alpha(K)\exp=0.019$ 1, K/L=2.8 6;
521.4 2	100	877.2	4 <sup>+</sup>	355.90	2 <sup>+</sup>	E2	0.0224	$\alpha(K)=0.01666$ 24; $\alpha(L)=0.00435$ 7; $\alpha(M)=0.001054$ 15; $\alpha(N+..)=0.000305$ 5 Mult.: supported by $\alpha(K)\exp=0.019$ 2, K/L=3.8, L/(M+)=2.7 15;
<sup>x</sup> 553.0 <sup>#</sup> 3	0.66 4	1430.1?	(5,6 <sup>+</sup> )	877.2	4 <sup>+</sup>			
<sup>x</sup> 566.4 4	0.3 1							
<sup>x</sup> 615.9 4	0.44 5							
633.5 3	1.15 5	2454.6	7 <sup>-</sup> ,8	1821.1	9 <sup>-</sup>			
647.3 2	95 3	2468.4	10 <sup>-</sup> ,11 <sup>-</sup>	1821.1	9 <sup>-</sup>	E2	0.01357	$\alpha(K)=0.01050$ 15; $\alpha(L)=0.00235$ 4; $\alpha(M)=0.000561$ 8; $\alpha(N+..)=0.0001628$ 23 Mult.: supported by $\alpha(K)\exp=0.0120$ 15, K/L=3.7 9.
<sup>x</sup> 659.5 <sup>@</sup>	<0.08							
<sup>x</sup> 673.9 2	0.18 4							
693.9 2	4.4 3	3162.4	(9 <sup>-</sup> ,10,11 <sup>-</sup> )	2468.4	10 <sup>-</sup> ,11 <sup>-</sup>			
722.0 <sup>#</sup> 4	0.67 7	3176.8?	(9 <sup>-</sup> )	2454.6	7 <sup>-</sup> ,8			
727.5 <sup>#</sup> 2	2.7 1	2889.3?	(9 <sup>-</sup> ,10,11 <sup>-</sup> )	2161.9?	(9 <sup>-</sup> ,10,11 <sup>-</sup> )			
760.6 <sup>#</sup> 3	0.78 5	3215.2?	(9 <sup>-</sup> )	2454.6	7 <sup>-</sup> ,8			
<sup>x</sup> 779.4 <sup>@</sup>	<0.04							
835.6 2	6.6 2	3304.0	(10,11 <sup>-</sup> )	2468.4	10 <sup>-</sup> ,11 <sup>-</sup>			
849.4 3	0.53 5	3304.0	(10,11 <sup>-</sup> )	2454.6	7 <sup>-</sup> ,8			

Continued on next page (footnotes at end of table)

$^{196}\text{Ir } \beta^-$  decay (1.40 h)    [1968Ja06](#) (continued) $\gamma(^{196}\text{Pt})$  (continued)

$E_\gamma$	$I_\gamma^{\ddagger\&}$	$E_i$ (level)	$J_i^\pi$	$E_f$	$J_f^\pi$	Mult. <sup>†</sup>	$\alpha^a$	Comments
<sup>x</sup> 868.1 3	0.48 4							
<sup>x</sup> 887.0 5	0.11 2							
<sup>x</sup> 893.0 5	0.20 2							
<sup>x</sup> 904.6 5	0.10 2							
914.6 <sup>#</sup> 3	0.30 5	1270.7	5 <sup>-</sup>	355.90	2 <sup>+</sup>	[E3]	0.01533	$\alpha(K)=0.01145$ 16; $\alpha(L)=0.00295$ 5; $\alpha(M)=0.000716$ 10; $\alpha(N+..)=0.000209$ 3
								$\alpha$ : E3 $\alpha$ (theory)'s mult. By 0.975 10 (Cf. <a href="#">1990Ne01</a> ).
<sup>x</sup> 926.0 5	0.07 2							
1024.6 <sup>#</sup> 3	0.26 3	2454.6	7 <sup>-</sup> ,8	1430.1? (5,6 <sup>+</sup> )				
1068 <sup>#</sup> 2	0.074 18	2889.3?	(9 <sup>-</sup> ,10,11 <sup>-</sup> )	1821.1	9 <sup>-</sup>			
1080.5 5	0.12 2	2454.6	7 <sup>-</sup> ,8	1374.0	7 <sup>-</sup>			
<sup>x</sup> 1091.5 <sup>@</sup>	<0.05							
<sup>x</sup> 1116.7 8	0.073 22							
<sup>x</sup> 1281.6 5	0.061 24							
1341.5 5	0.29 3	3162.4	(9 <sup>-</sup> ,10,11 <sup>-</sup> )	1821.1	9 <sup>-</sup>			
1355.8 <sup>#</sup> 5	0.06 1	3176.8?	(9 <sup>-</sup> )	1821.1	9 <sup>-</sup>			
1394.0 <sup>#</sup> 5	0.074 15	3215.2?	(9 <sup>-</sup> )	1821.1	9 <sup>-</sup>			
1482.5 4	2.4 2	3304.0	(10,11 <sup>-</sup> )	1821.1	9 <sup>-</sup>			

<sup>†</sup> From the adopted gammas.<sup>‡</sup>  $\gamma$  intensities normalized to  $I_\gamma(521)=100$  ([1968Ja06](#)).

# Placement based primarily on energy sum.

@ Unobserved in spectrum; limit placed on  $I_\gamma$ .

&amp; For absolute intensity per 100 decays, multiply by 0.96 3.

a Total theoretical internal conversion coefficients, calculated using the BrIcc code ([2008Ki07](#)) with Frozen orbital approximation based on  $\gamma$ -ray energies, assigned multipolarities, and mixing ratios, unless otherwise specified.<sup>x</sup>  $\gamma$  ray not placed in level scheme.

$^{196}\text{Ir} \beta^- \text{ decay (1.40 h)} \quad 1968\text{Ja06}$ 