

^{193}Ir IT decay (10.53 d) [2004Ni14](#),[1987Li16](#)

Type	Author	History	Citation	Literature Cutoff Date
Full Evaluation	M. Shamsuzzoha Basunia		NDS 143, 1 (2017)	31-Mar-2017

Parent: ^{193}Ir : E=80.238 6; $J^\pi=11/2^-$; $T_{1/2}=10.53$ d 4; %IT decay=100.0

 ^{193}Ir Levels

Data are from [1987Li16](#), except where noted, highly pure sources from $^{192}\text{Os}(n,\gamma)$ E=thermal, osmium enriched to 99.4% in ^{192}Os , chemical separation; measured E_γ , I_γ (absolute) (calibrated planar germanium, well-type germanium detectors), $E(\text{ce})$, $I(\text{ce})$ (absolute) ($4\pi\beta$ proportional counter, evacuated windowless Si(Li) detector); determined Ir L-subshell fluorescence yield and Coster-Kronig coefficients.

[2004Ni14](#) accurately measured the isomeric transition $\alpha(\text{K})$; enriched (99.935%) ^{192}Os target, chemical separation, HPGe detector.

E(level)	$J^\pi^\dagger$	$T_{1/2}$	Comments
0.0	$3/2^+$	stable	
80.238 6	$11/2^-$	10.53 d 4	%IT=100 $T_{1/2}$: Absolute electron counting in $4\pi\beta$ proportional counter (1987Li16). Other values: 11.9 d 5 (1957Bo12), 10.8 d 5 (1969Bi01), 12 d 2 (1970Ba56), 10.60 d 11 (1975Ba35).

† From Adopted Levels.

 $\gamma(^{193}\text{Ir})$

It has been suggested that $\alpha(\text{K})$ for this isomeric transition is decreased due to an ‘electronic bridge’ effect ([1988Zh11](#)); however, additional calculations ([1989Ba76](#),[1990Ba48](#)) seem to indicate that this second-order effect does not contribute measurably to the α of this transition. This effect is further discussed in [1989Pi14](#), [1990Ko06](#), [1990Ko22](#), [1990Ko28](#), [1992Tk01](#), [1994Tk02](#). [2004Ni14](#) gives precise value of measured $\alpha(\text{K})$ and compares the experimental value with several theoretical calculations.

E_γ	$E_i(\text{level})$	J^π_i	E_f	J^π_f	Mult.	$\alpha^\ddagger$	$I_{(\gamma+\text{ce})}^\dagger$	Comments
80.22 2	80.238	$11/2^-$	0.0	$3/2^+$	M4	2.14×10^4	100	ce(K)/($\gamma+\text{ce}$)=0.00503; ce(L)/($\gamma+\text{ce}$)=0.681; ce(M)/($\gamma+\text{ce}$)=0.235; ce(N+)/($\gamma+\text{ce}$)=0.0793 E_γ : other values: $E_\gamma=80.19$ 5 (1957Bo12), $E_\gamma=80.27$ 4 (1970Ba56). Other: 1966Sy01 . Mult.: from L1:L2:L3 (exp)=54.5 3:10.9 2:225 1. Also: $\alpha(\text{K})_{\text{exp}}=103.0$ 8 (from $I(\text{K x ray})/I_\gamma$ assuming $\omega(\text{K})=0.958$ 4 given by 1996Sc06) (2004Ni14). Other: $\alpha(\text{K})_{\text{exp}}=104$ 3 ($I(\text{K x ray})/I_\gamma$) (1987Li16); $\alpha(\text{K})_{\text{exp}}=92.6$ 9 ($I(\text{K x ray})/I_\gamma$) (1988Zh11). K:L:M:N:(O+P)(exp)= 2.08 4:290.4 6:101.4 6:26.6 3:3.8 1 (Ice(K) deduced from $I(\text{K x ray})$ (Ir fluorescence yield =0.95)); (O+P)/N3(exp)=0.228 2 (1975Ma32). Other ce data: 1970Ba56 , 1969Bi01 , 1957Bo12 .

† Absolute intensity per 100 decays.

‡ Total theoretical internal conversion coefficients, calculated using the BrIcc code ([2008Ki07](#)) with Frozen orbital approximation based on γ -ray energies, assigned multiplicities, and mixing ratios, unless otherwise specified.

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Decay Scheme

%IT=100.0

