

$^{130}\text{Cs IT decay (3.46 min)}$ **1983We07**

Type	Author	History Citation	Literature Cutoff Date
Full Evaluation	Balraj Singh	NDS 93,33 (2001)	11-May-2001

Parent: ^{130}Cs : E=163.25 *II*; $J^\pi=5^-$; $T_{1/2}=3.46$ min *6*; %IT decay=99.8 ^{130}Cs -%IT decay: %IT=99.84.1983We07: measured E γ , I γ , ce, $\gamma\gamma$. $^{130}\text{Cs Levels}$

E(level)	J^π [†]	$T_{1/2}$
0.0	1^+	
80.37 <i>7</i>	2^+	
131.54 <i>6</i>	2^+	
148.35 <i>7</i>	2^-	
163.25 <i>II</i>	5^-	3.46 min <i>6</i>

† From Adopted Levels.

 $\gamma(^{130}\text{Cs})$ I γ normalization: $\Sigma(I(\gamma+\text{ce}))$ of γ 's to g.s.=100. $\alpha(\text{exp})=\text{Ice}/I\gamma$ normalized to $\alpha(K)(148.35)=0.0631$ (E1 1968Ha53).

E_γ	I_γ &	$E_i(\text{level})$	J_i^π	E_f	J_f^π	Mult. [†]	α^a	$I_{(\gamma+ce)}$ &	Comments
14.9 [‡] <i>3</i>	0.000049 [#] <i>7</i>	163.25	5^-	148.35	2^-	[M3]	2.2×10^6 <i>3</i>	107 [#] <i>3</i>	$\alpha(L)=1.635 \times 10^6$; $\alpha(M)=4.26 \times 10^5$ $\text{ce}(M)<150$.
31.5 [‡] <i>3</i>	0.070 [#] <i>7</i>	163.25	5^-	131.54	2^+	[E3]	1.50×10^4 <i>9</i>	1050 [#] <i>80</i>	$\alpha(L)=1.141 \times 10^4$; $\alpha(M)=2.73 \times 10^3$ $\text{ce}(L) \approx 1030$, $\text{ce}(M) \approx 340$.
51.18 <i>5</i>	138 <i>6</i>	131.54	2^+	80.37	2^+	M1	6.34		$\alpha(K)\text{exp} \approx 3$; $\alpha(L)\text{exp} = 0.50$ <i>13</i> ; $\alpha(M)\text{exp} = 0.164$ <i>15</i> $\alpha(K) = 5.43$; $\alpha(L) = 0.726$; $\alpha(M) = 0.1481$; $\alpha(N..) = 0.0387$ $\text{ce}(K) \approx 570$, $\text{ce}(L) = 95$ <i>24</i> , $\text{ce}(M) = 31$ <i>2</i> . $\alpha(\text{exp}) = 6.6$ <i>6</i> from intensity balance.
80.45 <i>10</i>	680 <i>30</i>			80.37	2^+	0.0	1^+	M1	$\alpha(K)\text{exp} = 1.37$ <i>10</i> ; $\alpha(L)\text{exp} < 0.27$; $\alpha(M)\text{exp} = 0.041$ <i>10</i> $\alpha(K) = 1.459$; $\alpha(L) = 0.1941$; $\alpha(M) = 0.0395$; $\alpha(N..) = 0.01036$ $\text{ce}(K) = 1270$ <i>40</i> , $\text{ce}(L) < 250$, $\text{ce}(M) = 38$ <i>9</i> .
82.9 <i>1</i>	13.0 <i>7</i>	163.25	5^-	80.37	2^+	E3 [@]	59.4		$\alpha(K)\text{exp} = 11.7$ <i>11</i> ; $\alpha(L)\text{exp} = 41$ <i>4</i> ; $\alpha(M)\text{exp} = 10.7$ <i>11</i> $\alpha(K) = 11.43$; $\alpha(L) = 37.4$; $\alpha(M) = 8.45$; $\alpha(N..) = 2.136$ $\text{ce}(K) = 208$ <i>12</i> , $\text{ce}(L) = 722$ <i>13</i> , $\text{ce}(M) = 190$ <i>5</i> .

Continued on next page (footnotes at end of table)

^{130}Cs IT decay (3.46 min) 1983We07 (continued) $\gamma(^{130}\text{Cs})$ (continued)

E_γ	$I_\gamma^{\&}$	$E_i(\text{level})$	J_i^π	E_f	J_f^π	Mult. [†]	α^a	Comments
131.50 7	25.4 20	131.54	2 ⁺	0.0	1 ⁺	M1	0.420	$\alpha(K)\exp=0.37\ 3$; $\alpha(L)\exp<0.11$; $\alpha(M)\exp<0.03$ $\alpha(K)=0.360$; $\alpha(L)=0.0477$; $\alpha(M)=0.00971$; $\alpha(N+..)=0.00255$ $\text{ce}(K)=13.0\ 1$, $\text{ce}(L)<4.0$, $\text{ce}(M)<1.1$.
148.35 7	100 3	148.35	2 ⁻	0.0	1 ⁺	E1 [@]	0.0734	$\alpha(K)\exp=0.064$; $\alpha(L)\exp=0.0076\ 10$; $\alpha(M)\exp=0.0029\ 5$ $\alpha(K)=0.0631$; $\alpha(L)=0.00822$; $\alpha(M)=0.00166$; $\alpha(N+..)=0.00042$ $\text{ce}(K)=8.70\ 14$, $\text{ce}(L)=1.04\ 13$, $\text{ce}(M)=0.40\ 6$.

[†] From ce data, except as noted.[‡] From conversion electron data only.[#] From $I(\gamma+\text{ce})$ and α ; $I(\gamma+\text{ce})$ is from intensity balance.[@] From K/L.[&] For absolute intensity per 100 decays, multiply by 0.051 2.^a Total theoretical internal conversion coefficients, calculated using the BrIcc code (2008Ki07) with Frozen orbital approximation based on γ -ray energies, assigned multipolarities, and mixing ratios, unless otherwise specified.

