

¹⁸⁰Os

The discovery of ¹⁸⁰Os was reported in the 1967 paper “The decay of the isotope Os¹⁸⁰” by Belyaev et al. (1966Be41). Tm₂O₃ targets were bombarded with 100 MeV ¹⁴N from the Dubna U-150 cyclotron and ¹⁸⁰Os was formed in the fusion-evaporation reaction ¹⁶⁹Tm(¹⁵N,4n). Gamma-ray spectra were measured with a NaI(Tl) crystal following chemical separation. “In our measurements we observed γ lines having the same energy, 105 and 510 keV, and an intense 880-keV γ line. This gives grounds for assuming that the observed 21-minute activity is connected with the decay of the isotope Os¹⁸⁰, which, taking into account the time necessary to separate the osmium (~ 1 hour), is in radioactive equilibrium with the daughter rhenium, that is, Os¹⁸⁰ $\xrightarrow{21 \text{ min}}$ Re¹⁸⁰ $\xrightarrow{2.4 \text{ min}}$.” Previously a 23 min half-life had been tentatively assigned to ¹⁸¹Os (1958Fo47) and was later reassigned to ¹⁸⁰Os (1966Ho16).

Adapted from reference (2012Ro36)

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