

⁵⁷Fe

In 1935, Aston discovered ⁵⁷Fe at the Cavendish Laboratory in Cambridge, UK, and described the results in his article “The Isotopic Constitution and Atomic Weights of Hafnium, Thorium, Rhodium, Titanium, Zirconium, Calcium, Gallium, Silver, Carbon, Nickel, Cadmium, Iron and Indium” ([1935As01](#)). Aston used a pure sample of the carbonyl in the spectrograph. “In addition to the strong isotope 56 and a weak one, 54, previously known, a third, 57 was revealed.”

Adapted from reference ([2010Sc18](#))

[1935As01](#) F. W. Aston, Proc. Roy. Soc. (London) **149**, 396 (1935).

[2010Sc18](#) A. Schuh, A. Fritsch, M. Heim, A. Shore, and M. Thoennessen, At. Data Nucl. Data Tables **96**, 817 (2010).

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