

^{32}S

Aston discovered ^{32}S at the Cavendish Laboratory in Cambridge, UK, in 1920 as reported in “The constitution of the elements” ([1920As02](#)). Isotopes were identified by measuring their mass spectra. “Sulphur (atomic weight 32.06) has a predominant constituent 32. Owing to possible hydrogen compounds the data are as yet insufficient to give a decision as to the presence of small quantities of isotopes of higher mass suggested by the atomic weight.”

Adapted from reference ([2012Th10](#))

[1920As02](#) F. W. Aston, *Nature* **105**, 547 (1920).

[2012Th10](#) M. Thoennessen, *At. Data Nucl. Data Tables* **98**, 933 (2012).

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