

Use of K_0 data in decay scheme normalization: An example



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Decay scheme normalization

- From decay scheme
- Absolute intensity of one γ -ray is known
- x-ray intensity is known
- Annihilation radiation intensity is known

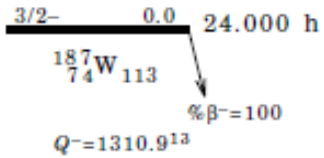
%intensity of γ -rays

Isotope	Half-life	γ -ray (keV)	TOI 7 th edition - '78	TORadio Iso '86	Nudat	year
⁵² V	3.75 m	1434.2	100	100 (1)	100.0 (14)	2007
²⁴ Na	15.02 h	1368.6	100	100	99.9936 (15)	2007
⁵¹ Ti	5.75 m	320.1	93.4 (9)	93.0 (4)	93.1 (4)	2006
³⁸ Cl	37.2 m	1642.4	31.0 (15)	31.0 (4)	33.3 (7)	2008
²⁸ Al	2.24 m	1778.7	100	100	100	1998
⁴⁹ Ca	8.72 m	3084.4	92 (1)	92.1 (10)	90.72 (4)	2008
⁴² K	12.36 h	1524.6	18.8 (6)	18.8 (6)	18.08 (9)	2001
⁶⁶ Cu	5.10 m	1039.2	8 (1)	7.4 (1.8)	9.23 (9)	2010
⁵⁶ Mn	2.58 h	1810.7	27.2 (8)	27.2 (7)	26.9 (4)	2011
^{60m} Co	10.5 m	58.6	2.0 (1)	2.01 (6)	2.0359 (6)	2003
¹⁸⁷ W	23.9 h	685	26.3 (2)	26.4 (6)	33.2 (5)	2009
⁷⁶ As	1.1 d	559.1	45 (2)	45 (2)	45 (2)	1995
¹¹⁰ Ag	24.6 s	657.7	4.5 (2)	4.5 (2)	4.50 (24)	2012
^{116m} In	54.15 m	416.9	32.4 (15)	29.2 (14)	27.2 (4)	2010
⁸² Br	1.47 d	776.5	83.4 (9)	83.6 (8)	83.4 (12)	2003
¹²⁸ I	25 m	442.9	16 (2)	16.9 (17)	12.62 (16)	2001
¹²² Sb	2.7 d	563.9	70 (4)	70.0 (4)	70.67 (18)	2007
^{69m} Zn	13.76 h	438.6	94.8 (3)	94.8 (3)	94.77 (20)	2000
^{125m} Sn	9.52 m	332.1	97.0 (4)	97.0 (19)	97.3 (14)	2011
²⁹ Al	6.5 m	1273	89.1 (7)	91.3	91.26 (6)	2012
^{77m} Se	17.4 s	162	52.5 (12)	52.4 (12)	53.2 (7)	2012

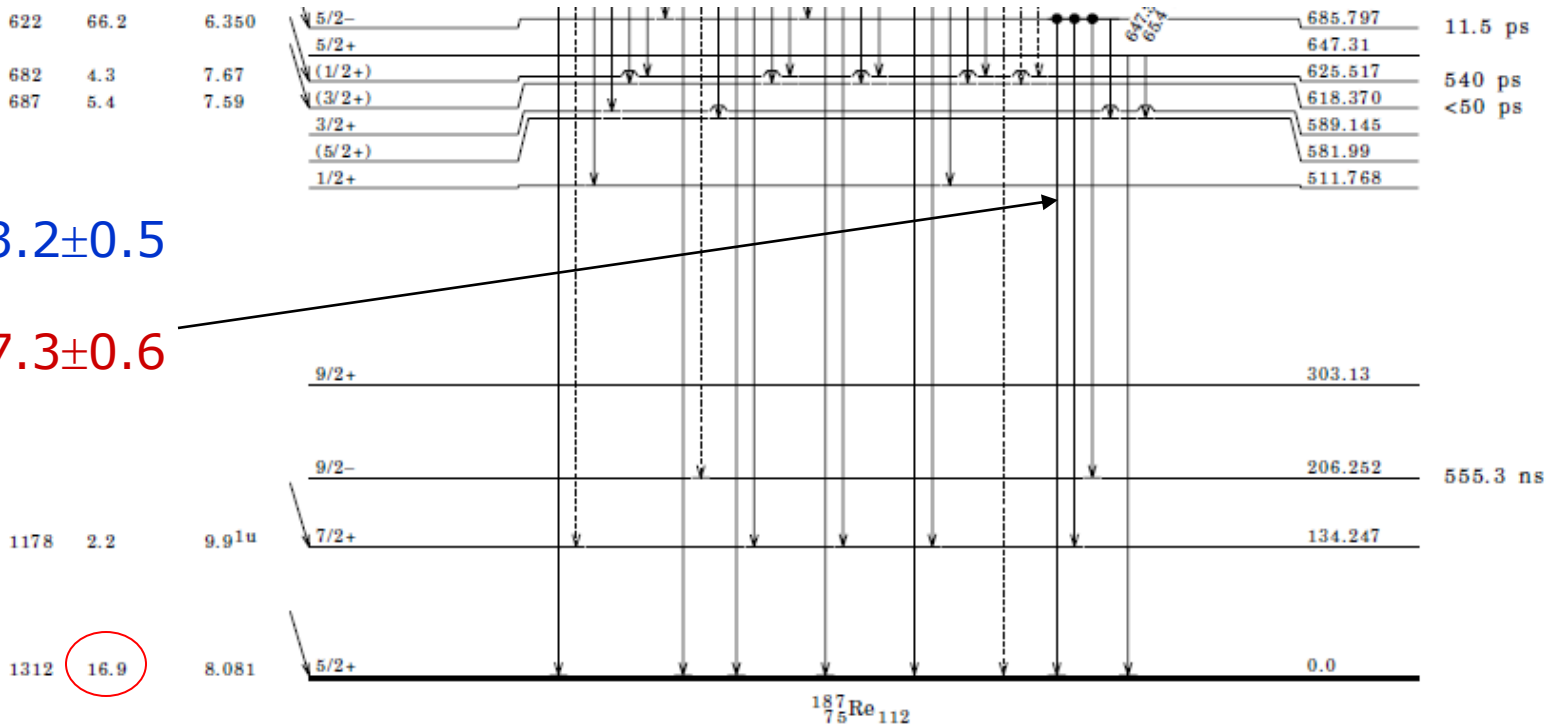
Decay Scheme (continued)

Intensities: I(γ +ce) per 100 parent decays
& Multiply placed; undivided intensity given

Two-Dimensional $4\pi\beta\text{-}\gamma$ Coincidence System
J.Nucl.Sci.Technol.(Tokyo) 36, 1119 (1999)
N. Marnada, *et al.* (1999Mb04)



E_{β^-}	I_{β^-}	Log ft	Transition	Energy (keV)
	0.041	6.75	$(3/2^+, 5/2^+)$	1230.12
	0.015	7.71	$(5/2^+)$	1190.46



$\%P_{\gamma} = 33.2 \pm 0.5$

$\%P_{\gamma} = 27.3 \pm 0.6$

Reliability of Prompt γ -Ray Intensities for the Measurement of Neutron Capture Cross Sections

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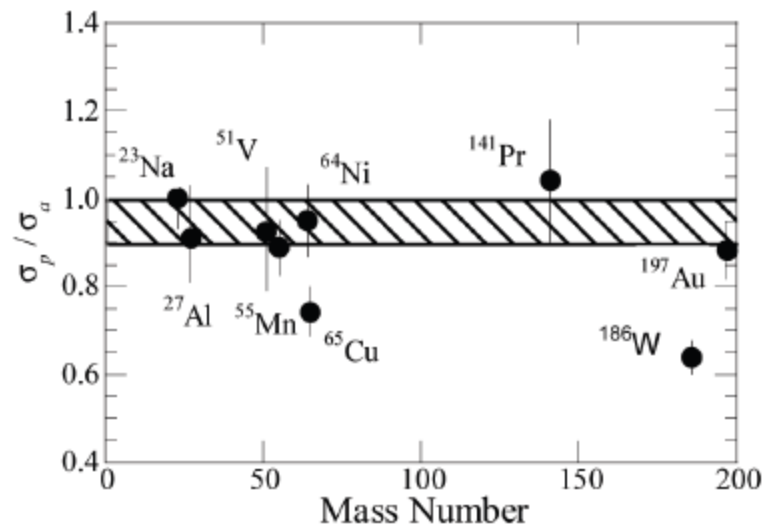


FIGURE 4. The ratio of σ used prompt γ -rays to σ used β -decay γ -rays. The detector used was the 22% HPGe detector. It is clearly seen that the ratios are lower systematically than 1.0 ($\sigma_{\text{prompt}} = \sigma_{\text{decay}}$).

Neutron Activation Analysis and K_0 -data

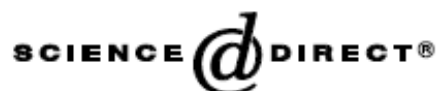
- Neutron activation analysis is a non-destructive method for elemental analysis in the sample
- K_0 -method eliminates detailed analytical steps by determining a constant for each elements
- Simultaneous irradiation of element with Au target and 412 keV delayed gamma-ray is used from the $^{197}\text{Au}(n,\gamma)^{198}\text{Au}$ reaction to determine

$$k_{0,\text{Au}}(\mathbf{a}) = \frac{M_{\text{Au}} \theta_{\mathbf{a}} \sigma_{0,\mathbf{a}} \gamma_{\mathbf{a}}}{M_{\mathbf{a}} \theta_{\text{Au}} \sigma_{0,\text{Au}} \gamma_{\text{Au}}}$$



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Atomic Data and Nuclear Data Tables 85 (2003) 47–67

Atomic Data
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Recommended nuclear data for use in the k_0 standardization of neutron activation analysis^{☆,☆☆}

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Abstract

k_0 factors (composite nuclear constants) for use in the k_0 standardization of reactor neutron activation analysis were experimentally measured with great care in several laboratories. The recommended values thus obtained for the relevant gamma rays of 144 analytically interesting radionuclides are tabulated, together with evaluated values for their associated parameters such as resonance integral to thermal cross-section ratios and effective resonance energies. A classification is also given of the various activation-decay types, to which the data are strictly correlated.

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Table 1

 $k_{0,Au}$ factors and related nuclear data for use in k_0 standardized neutron activation analysis. See page 52 for Explanation of Tables

Element	Target isotope	Q_0 (s, %)	\bar{E}_T , eV	Formed isotope (Activation/decay code)	$T_{1/2}$	E_γ , keV	Measured $k_{0,Au}$ (s, %)	Notes
F	¹⁹ F	2.2	44,700	²⁰ F (I)	11.16 s	1633.6	9.98E-4 (1.2)	Q_0 adopted
Na	²³ Na	0.59 ()	3380	²⁴ Na (IVb)	14.96 h	1368.6 2754.0	4.68E-2 (0.6) 4.62E-2 (0.9)	Data for m + g (m 20.2 ms)
Mg	²⁶ Mg	0.64	257,000	²⁷ Mg (I)	9.462 min	170.7 843.8 1014.4	3.02E-6 (1.0) 2.53E-4 (0.5) 9.80E-5 (0.2)	Q_0 adopted
Al	²⁷ Al	0.71	11,800	²⁸ Al (I)	2.2414 min	1778.9	1.75E-2 (0.8)	Q_0 adopted
Si	³⁰ Si	1.11 (6.)	2280	³¹ Si (I)	2.622 h	1266.2	1.45E-7 (0.7)	
S	³⁶ S	1.12 ()		³⁷ S (I)	5.05 min	3103.4	1.96E-6 (1.8)	No resonance data; large terrestrial variation in θ
Cl	³⁷ Cl	0.69 ()	13,700	³⁸ Cl (IVb)	37.24 min	1642.7 2167.4	1.97E-3 (1.4) 2.66E-3 (1.3)	Data for m + g (m 715 ms)
Ar	⁴⁰ Ar	0.63	31,000	⁴¹ Ar (I)	1.822 h	1293.6	3.32E-2 ()	Q_0 adopted
K	⁴¹ K	0.87 (3.)	2960	⁴² K (I)	12.36 h	312.7 1524.7	1.59E-5 (1.1) 9.46E-4 (0.6)	
Ca	⁴⁶ Ca	1.3		⁴⁷ Ca (I) ↓ ⁴⁷ Sc (IIa)	4.536 d 3.349 d	489.2 807.9 1297.1 159.4	9.14E-8 (1.8) 9.20E-8 (0.2) 9.54E-7 (1.7) 8.57E-7 (1.6)	No resonance data; Q_0 adopted; Large terrestrial variation in θ
	⁴⁸ Ca	0.45	1,330,000	⁴⁹ Ca	8.718 min	3084.4	1.01E-4 (0.9)	Q_0 adopted

➤ Corrections and new values in NIM A 622 (2010) 377–380

% intensity of 685 keV γ -ray of ^{187}W

$$\gamma_a = \frac{k_{0,Au} M_a \theta_{Au} \sigma_{0,Au} \gamma_{Au}}{M_{Au} \theta_a \sigma_{0,a}}$$

- From K_0 and Mughabghab value: 32 (4)
 - using 38.1 (5) barn for (n,γ)
- Latest evaluation 2009: 33.2 (5)
- Earlier evaluation 1991: 27.3 (6)

J. Rad. Nucl. Chem., vol. 257, 2003, p493 for cross-section

Thank you

